Chem 220

Statistics for Calibration Methods

Linear Least Squares Method for *n* points (x_i, y_i) fitted to line y = mx + b

slope =
$$m = \frac{n\sum(x_iy_i) - \sum x_i\sum y_i}{D}$$

y-intercept = $b = \frac{\sum(x_i^2)\sum y_i - \sum(x_iy_i)\sum x_i}{D}$
where $D = n\sum(x_i^2) - (\sum x_i)^2$

Assuming errors in y are larger than errors in x (known amounts of standards), the uncertainty in the y values is

$$s_{\mathcal{Y}} = \sqrt{\frac{\sum (y_i - mx_i - b)^2}{n - 2}}$$

Least squares minimizes the numerator: (vertical deviations of *y* from the line)². Notice the n - 2 degress of freedom, where *n* is the number of points.

The uncertainty in m and b come from the uncertainty in y.

standard deviation of slope =
$$s_m = s_y \sqrt{\frac{n}{D}}$$

standard deviation of intercept = $s_b = s_y \sqrt{\frac{\sum (x_{i2})}{D}}$

If use a calibration curve to find x from known y, then standard deviation for x is

$$s_{\mathcal{X}} = \frac{s_{\mathcal{Y}}}{|m|} \sqrt{\frac{1}{k}} + \frac{n\overline{x}^2 + \sum(x_i^2) - 2\overline{x}\sum x_i}{D}$$

where \overline{x} is the average value calculated from y = mx + b, k is the number of measurements averaged, and n is the number of points in the calibration curve. The confidence intervals will be

$$m \pm t s_m$$
 $b \pm t s_b$ $x \pm t s_x$

where the *t*-value depends on the confidence level and n - 2 degrees of freedom.

Standard Addition Method Extrapolation

xextrapolated =
$$\frac{b}{m}$$
 $s_{xextrapolated} = \frac{s_y}{|m|} \sqrt{\frac{1}{n} + \frac{\overline{y}^2}{m^2 \sum (x_i - \overline{x})^2}}$

where \overline{x} and \overline{y} are the averages of the *x* and *y*-values and *n* is the number of points in the standard addition curve (including the point with no standard added). The confidence intervals will be

$$x \pm t \; s_{x_{extrapolated}}$$

where the *t*-value depends on the confidence level and n - 2 degrees of freedom.